



CHEMISTRY

0620/63

Paper 6 Alternative to Practical

May/June 2017

MARK SCHEME

Maximum Mark: 40

Published

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This document consists of **4** printed pages.

Question	Answer	Marks
1(a)(i)	(delivery) tube	1
1(a)(iii)	arrow beneath the tube containing the mixture of alcohols	1
1(b)	to cool	1
	the gas into a liquid	1
1(c)	to measure the temperature of the vapour / temperature of liquid would not be constant	1
1(d)	E shown on the test-tube in water bath	1
1(e)(i)	lighted splint ignites the liquid / test for water, e.g. add anhydrous copper(II) sulfate gives a negative result	1
1(e)(ii)	melting / boiling point determination	1

Question	Answer	Marks
2(a)	all volume boxes completed correctly: 0, 13, 25, 38, 48, 59, 70, 79, 88, 96	3
2(b)	origin plotted	1
	other points correctly plotted	1
	two smooth lines	1
	labelled	1
2(c)	Experiment 1	1
	more concentrated / stronger acid / the acid has a lower pH	1

Question	Answer	Marks
2(d)	volume of gas at 30 s	1
	correct calculation of rate	1
	unit: cm^3/s OR cm^3s^{-1}	1
2(e)	all the magnesium will have reacted	1
2(f)	faster reaction / increased rate	1
	magnesium powder has a higher surface area	1
2(g)	advantage: easy to use / quick	1
	disadvantage: not accurate	1
2(h)	use of burette / pipette / gas syringe / weighed amount of magnesium / repeat experiment (and average) / clean the magnesium / remove oxide layer	1

Question	Answer	Marks
3(a)	chlorine	1
3(b)(i)	iron(III)	1
	hydroxide	1
3(b)(ii)	green	1
	precipitate	1
3(c)	oxygen	1
3(d)	catalyst	1
	transition element compound / manganese oxide	1

Question	Answer	Marks
4	any 6 from: crush lumps pestle and mortar weigh cassiterite heat/reduce with carbon/CO/more reactive metal, e.g. Zn weigh tin $(\text{mass of tin} / \text{initial mass}) \cdot 100 (\%)$	6